

OBITUARY

S.G. Whiteway (1927-2001)

Dr. Stirling G. Whiteway, who died 23 October 2001, was born in 1927 in Stirling, Nova Scotia. In 1944 he won the Hector McInnes Memorial Scholarship to read science at Dalhousie University, graduating in 1947 after majoring in chemistry. The following year he obtained a National Research Council (NRC) post-graduate bursary and a diploma in chemical engineering. The bursary was extended to 1949 during which he earned the degree of M.Sc. in physical chemistry. He then won a National Research Council overseas science research scholarship to be held under E.K. Rideal at the Royal Institution in London, U.K. but shortly afterwards Rideal moved to Cambridge and Whiteway returned to Canada to work with S.G. Mason at McGill University, where he obtained a National Research Council Fellowship which supported his studies for the degree of Ph.D. He moved to N.R.C. in Ottawa in 1952 to take up a post-doctoral fellow appointment. Shortly afterwards, in 1953, he returned to Nova Scotia to join the National Research Council Laboratory in Halifax. He became Principal Research Officer and was Head of the Industrial Materials Section which managed national programs in coal and ceramic research and included collaboration with industries such as Sydney Steel. He retired from NRC in 1989. Dr. Whiteway was elected to membership in the Nova Scotian Institute of Science (NSIS) in 1947 and was a Life Member. In addition he was a fellow of the Canadian Institute of Chemistry and a member of many other societies including the Canadian Institute of Mining, Metallurgy and Petroleum, American Ceramic Society, Canadian Ceramic Society and Sigma Xi (The Scientific Research Society). He served on Natural Sciences and Engineering Research Council granting committees for six years.

Originally appointed an adjunct Professor at the Technical University of Nova Scotia (TUNS) in 1985 in the Department of Mining and Metallurgical Engineering, he was reappointed in 1990 for consecutive three-year periods, most recently in the Faculty of Engineering (formerly TUNS), Dalhousie University. During this period he taught undergraduate and graduate courses and supervised numerous undergraduate theses.

His first contribution (in association with Professor C.C. Coffin and others) to the NSIS occurred in December 1947 when he demonstrated the experimental measurement of ^{32}P (half-life 14 d, β -1.71 Mev). The technique was used (Proc. N.S. Inst. Sci., 22(3):42) to measure the uptake of the radionuclide by organisms in an acid bog-lake and has been used by molecular biologists ever since. Also under Coffin, in a similar vein, he studied the diffusion of silver nitrate and Ag^+ in aqueous solution using $^{110}\text{Ag}^+$ (250 d. β 0.53 Mev, γ 0.9 Mev) as a marker for the cation. This experimental work reveals two of Whiteway's extraordinary skills – his ability to handle safely very hazardous materials and a numerical and algebraic talent – both of which were manifest throughout his career.

His work with Mason at McGill University solved many discrepant results in the literature on the liquid vapour coexistence curves of ethane, ethylene and xenon. It was shown that these discrepancies were mostly due to impurities and agitation. There was therefore good agreement that a major contributor to the "flat-top effect" was gravitational compression.

The very productive collaboration between C.R. Masson and Whiteway in photochemistry, studies of gas-metal systems e.g. the use of levitation to determine the solubility of oxygen in iron at 1800-1900° and the development of a theory of the

proportion of oligomers in polymerization reactions were described in Whiteway's contribution to the obituary of C.R. Masson (Proc. N.S. Inst. Sci., 38:189-197), with one notable exception – Whiteway's development of Pascal's triangle for calculating the probabilities of such size distributions. This early example of fuzzy logic (Furuhashi and Uchikawa, 1995) underpinned the intellectual controversies stimulated by Masson, and led to a greater understanding of the chemistry of metallurgical slags.

Whiteway's commitment to public service was exemplified by his contribution to the effort in 1970 by many Canadian scientists and engineers to understand and mitigate the effects of a major spill of "Bunker C" oil into Chedabucto Bay, Nova Scotia.

In the late 1970s Whiteway's interests tended to concentrate on the chemistry of the transformation of coal and iron ore into steel. His description of this chemistry (Proc. N.S. Inst. Sci., 30:67-87) is an authoritative and lucid contribution to our knowledge of the subject and, indeed, the status of the industry in 1980. It would be presumptuous to attempt to summarize this account because it does not necessarily draw attention to the ingenious solutions to the problems of studying chemistry on a laboratory scale at temperatures in the region 1000-2000°. The use of levitation has been mentioned, but not its application to the measurement of the activity coefficients of ferrous oxide in equilibrated iron-ferrous oxide-silicon dioxide melts at $\approx 1900^\circ$. Another notable contribution to high temperature chemistry was the fabrication of ceramic vessels by slip-casting techniques, and studies of their corrosion at elevated temperatures, resistance to thermal shock, abrasion and the relationship between particle size (studied by scanning electron microscopy) and their development of microcracks.

In Whiteway's review of the steel industry he mentions the discovery of transition metal sulfides in coal seams in Cape Breton. Work on these particles ($\approx 20 \mu\text{m}$) became part of a general investigation of the changes occurring in the conversion of coal into coke (char). These investigations included the effect of temperature and reduced pressure on the process. The use of "free fall" stainless steel (600 cm high) and quartz (140 cm) reactors was used to determine the production under different conditions of hydrogen, carbon oxides and hydrocarbons. The very complex chemistry of the decomposition of the iron sulfide occlusions was not described in detail. It was shown that the crystalline material (marcasite) was more stable at $\approx 500^\circ$ than the pyritic amorphous material and that the sulfur liberated was found partly in the coal tar.

Towards the end of his career, Whiteway returned to problems of diffusion, in particular the diffusion of hydrogen in steel. The subject is of theoretical interest and industrial importance. For example, the mechanical integrity of steel rail used in railroad construction is dependent in part on the hydrogen content of the steel developed during its manufacture. Whiteway and his collaborators constructed an apparatus, fabricated in part from steel rail produced in the mill in Sydney, Nova Scotia, which permitted the study of hydrogen diffusion and "solubility" in steels. It was found for example that the solubility of hydrogen in a steel containing 0.7% carbon, 0.8% manganese and 0.5% silicon at 1000° was $1.9 \mu\text{g cm}^{-3}$ and at 770° $0.5 \mu\text{g cm}^{-3}$, and that the diffusivity over the same temperature range of the same sample was virtually unchanged at $10^{-3.7} \text{ cm}^2 \text{ s}^{-1}$. The figures are given to illustrate the sensitivity and precision of the method which was materially assisted by a general solution of the partial differential equations involved, by Whiteway's son. Whiteway notes in these publications that "the diffusivity of an atom in a solid is a fundamental property which reflects the mobility of the atom and the structure of the host matrix". This fundamental property has become of increasing importance in many fields of chemistry from the diffusion of ions through nafion membranes in the industrial manufacture of caustic soda (Vaughan, 1973; Ramp, 1976), to the migration of activated amino acids in

channels of polyribosyl phosphates (Nissen et al., 2000). We must hope that in Pasternak's beautiful words: "(For) others will surely follow/The living imprint of your feet".

For almost 50 years Stirling Whiteway and his wife provided fine hospitality, encouragement and help to many students, post-doctorate fellows and friends, many of whom are now dispersed in different parts of the world. An example of this hospitality and other interests may be found in his last publication in these Proceedings (Proc. N.S. Inst. Sci., 40:59-60). We are proud to have known him and offer our sincere condolences to his wife and family.

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